

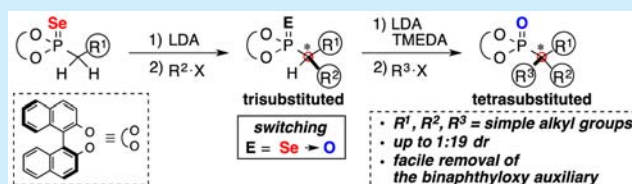
Sequential Deprotonation–Alkylation of Binaphthyloxy-Substituted Phosphonochalcogenoates: Chiral Tri- and Tetrasubstituted Carbon Centers Adjacent to a Phosphorus Atom

Yuuki Maekawa, Toshifumi Maruyama, and Toshiaki Murai*

Department of Chemistry and Biomolecular Science, Faculty of Engineering, Gifu University, Yanagido, Gifu 501-1193, Japan

Supporting Information

ABSTRACT: Sequential deprotonation and alkylation of 1,1'-binaphthyloxy-substituted phosphoselenoates and phosphonates resulted in the diastereoselective formation of chiral tri- and tetrasubstituted carbon centers adjacent to a phosphorus atom.

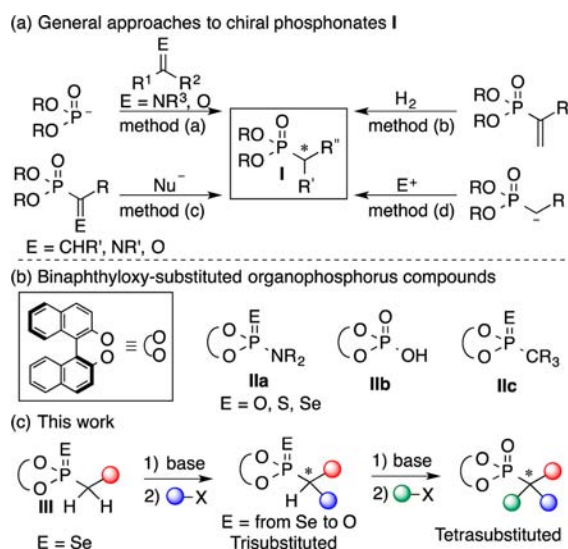


Optically active phosphonic acids and their derivatives (I) with chiral carbon centers adjacent to the phosphorus atom are not only biologically relevant¹ but also versatile synthetic intermediates toward optically active organophosphorus compounds.² Accordingly, several synthetic methods for the formation of I with trisubstituted carbon atoms have been developed in past decades^{3–5} involving the asymmetric addition of phosphorus anions to unsaturated compounds such as carbonyl compounds and imines (Scheme 1a, method a),³ the asymmetric reduction of vinyl phosphonates (Scheme 1a, method b),⁴ the asymmetric addition of carbon nucleophiles to unsaturated carbon atoms adjacent to the phosphorus atom (Scheme 1a, method c),⁵ and the addition of carbanions adjacent to the phosphorus atom to electrophiles with chiral ligands (Scheme 1a, method d).⁶ Alternatively, the use of chiral

auxiliaries is a reliable method to control the stereochemistry of the reaction of carbanions, whereby the judicious choice of chiral auxiliary is crucial. In fact, optically active 1,2-amino-alcohol,⁷ 1,2-diamine,⁸ and menthyl-tethered⁹ phosphonates have been used as precursors for chiral organophosphorus compounds such as I. However, even with these established methods, the construction of chiral centers substituted with only simple alkyl groups next to the phosphorus atom remains a challenging task. Meanwhile, a large variety of synthetic reactions involving organophosphorus compounds with binaphthyloxy groups, such as phosphoric amides IIa¹⁰ and phosphoric acids IIb,¹¹ have been developed over the past decades (Scheme 1b). In contrast, substantially less attention has been paid to phosphonates such as IIc,¹² despite the fact that the binaphthyloxy group in IIc exerts a strong influence on the stereochemical fate of the carbon atom connected to the phosphorus atom (Scheme 1a, method d). Indeed, our recent studies showed that the chirality of the α carbon atoms relative to the phosphorus atom in phosphonochalcogenoates III can be discriminated by ³¹P NMR spectroscopy.¹³ Moreover, carbanions at these positions may exhibit chirally distinct environments. Herein, we report the sequential deprotonation–alkylation of phosphonochalcogenoates III (Scheme 1c). Changing the elements attached to the phosphorus atom from selenium to oxygen in III afforded control over highly efficient diastereoselective reactions leading to tri- and tetrasubstituted chiral carbon centers.

Initially, we treated *P*-phenethyl phosphonate 1a consecutively with a base and methyl iodide (MeI) as an electrophile (Table 1, entry 1). Several bases including lithium diisopropylamide (LDA) were tested at –70 °C, followed by the addition of MeI, which resulted in the formation of 4a and 7a in 49 and 27% yield, respectively. To suppress the formation of 7a, which is probably formed via the dimethylation of 1a, the varying

Scheme 1. Synthetic Routes to Chiral Phosphonate Derivatives



Received: August 29, 2016

Published: September 29, 2016

Table 1. Successive Deprotonation–Methylation of Phosphonates and Phosphonochalcogenoates^a

entry	E	4a–6a yield (%)	dr	7a–9a yield (%)
1	O (1a)	49 ^b (4a)	10:90	27 ^b (7a)
2	S (2a)	85 (5a)	8:92	0 (8a)
3	Se (3a)	89 (6a)	5:95	0 (9a)

^aIsolated yield; dr determined by ³¹P NMR analysis of the crude reaction mixtures. ^bYield determined by the NMR analysis.

conditions for the reaction between 1a and LDA were examined. Unfortunately, these attempts remained unsuccessful. Under the aforementioned conditions, *P*-phenethyl phosphonothioate 2a and -selenoate 3a¹⁴ furnished an increased yield of 5a and 6a, respectively, and the corresponding dimethylated products were not observed (entries 2 and 3). It is particularly noteworthy that the reaction of phosphoselenoate 3a afforded the highest diastereoselectivity and yield among 1a–3a.¹⁵

With these results in hand, we examined the sequential reaction of 3a with LDA and a range of electrophiles (Table 2).

Table 2. Reaction Scope with Respect to Electrophiles^a

EX	product isolated yield (%) ^b
<i>n</i> -C ₆ H ₁₃ I	6b (75 (5>95 dr))
ArCH ₂ Cl	6c (71 (5:95 dr))
Ar = 4-MeOC ₆ H ₄	
ArCH ₂ Br	6d (74 (5>95 dr))
Ar = 4-BrC ₆ H ₄	
BrCH ₂ CH=CH ₂	6e (70 (7: 93 dr))
ArC(O)Cl	6f (45 (15:85 dr))
Ar = 4-CF ₃ C ₆ H ₄	
	6g (74 (10:90 dr))
Ar = 4-MeOC ₆ H ₄	
Me ₃ SiCl	6h (94 (5:95 dr))
Bu ₃ SnCl	6i (57 (5>95 dr))

^aTwo procedures were employed for these reactions, see [Supporting Information](#). ^bIsolated yield; dr determined by ³¹P NMR analysis of the crude reaction mixtures.

Under the previously established optimal conditions, the reaction proceeded with various halides, affording the corresponding products (6b–6i) in high diastereoselectivity and good to high yield. The use of benzyl halides bearing electron-withdrawing and -donating groups on an aromatic ring furnished 6c and 6d in high diastereoselectivity. Notably, the chirality of the α carbon atom in 6c and 6d originates from a remote position, that is, from the different substituents at the *para*-positions of the aromatic ring. Nevertheless, during the reaction, those differences were clearly discriminated. α -Acylated products 6f and 6g were prepared in good diastereoselectivity from the

reaction with acid halides, containing electron-withdrawing and -donating groups on an aromatic ring. Furthermore, the α -functionalization of 3a with Me₃SiCl and Bu₃SnCl also proceeded smoothly to generate 6h and 6i in high diastereoselectivity.

Subsequently, we explored the scope of this reaction with respect to phosphoselenoates (Table 3). Even though the

Table 3. Reaction Scope with Respect to Phosphoselenoates^a

reactant	EX	product	yield (%) ^b
10	PhCH ₂ Br	6a	84 (50:50 dr)
11	PhCH ₂ Br	17	71 ^c (73:27 dr)
12	MeI	18	58 (10:90 dr)
13	MeI	19	87 (90:10 dr)
14	MeI	20	83 (93:7 dr)
15	MeI	21	74 ^c (93:0:7:0 dr)
16	PhCH ₂ Br	6c'	46 (95:5 dr)

^aTwo procedures were employed for these reactions; see [Supporting Information](#). ^bIsolated yield; dr determined by ³¹P NMR analysis of the crude reaction mixtures. ^cReaction conducted at –40 °C. ^dYield and dr after recrystallization are shown in parentheses.

successive reaction of *P*-ethyl phosphoselenoate 10 with LDA and benzyl bromide proceeded smoothly at –70 °C and afforded 6a in high yield, no diastereoselectivity was observed. Sterically hindered ester 11, containing a *tert*-butyl group, also generated the corresponding carbanion at –40 °C, and the alkylation of the resulting carbanion furnished 17 in good diastereoselectivity. Surprisingly, when ester 12, bearing a long-chain alkyl group, was used as a substrate, methylated 18 was also obtained in high diastereoselectivity. Under the applied reaction conditions, silyl and acetal groups were tolerated well, and esters 19 and 20 were obtained in high diastereoselectivity.

Methylation of ester 15, bearing one chiral carbon center, afforded ester 21, which contains two adjacent chiral carbon centers at the α and β position relative to the phosphorus atom, in high diastereomeric excess. The reaction of ester 16 with benzyl bromide also proceeded with high efficiency and diastereoselectivity to furnish diastereoisomer 6c' exclusively. The stereochemistry of 6c' is antipodal to that of 6c, which is derived from ester 3a. This result suggests that the stereochemistry of the reaction of the anions generated from the phosphoselenoates and electrophiles is controlled kinetically.

In some cases, the obtained diastereomers could be separated by ordinary recrystallization or column chromatography on silica gel.

The absolute configurations of **6a**, **6c**, and **21** were unequivocally determined by single-crystal X-ray diffraction analysis.¹⁶

To construct rare examples of chiral tetrasubstituted carbon centers,¹⁷ in particular, with three carbon-containing substituents,¹⁸ adjacent to a phosphorus atom, we then attempted the lithiation of ester **6h** (Table 4). However, under a variety of

Table 4. Construction of Tetrasubstituted Carbon Atoms Adjacent to a Phosphorus Atom^a

entry	E	<i>t</i> (°C)	yield (%)	dr
1	Se (6h)	−70	nr (22)	
2	O (4h)	−70	nr (23)	
3	O (4h)	−40	59 (23)	5:>95
4 ^b	O (4h)	−40	80 (23)	7:93

^aIsolated yield; dr determined by ³¹P NMR analysis of the crude reaction mixtures. ^bTMEDA (3.3 equiv) was used as an additive.

reaction conditions employing other bases and increasing the reaction temperature, the lithiation was unsuccessful and **6h** remained mostly intact. Based on the results in Table 1, we subsequently focused on phosphonate **4h**, as the deprotonation may occur at the carbon atom adjacent to the phosphorus atom in **4h**. As expected, the lithiation of phosphonate **4h** at −40 °C, followed by the addition of MeI, afforded **23** in high diastereoselectivity (Table 4, entry 3). The use of TMEDA (1.6 equiv) as an additive enhanced the yield of **23** (Table 4, entry 4).

Under optimized reaction conditions, phosphonates with a trisubstituted carbon atom at the α position relative to the phosphorus atom also engaged in this reaction sequence (Table 5). The lithiation–alkylation of phosphonates containing aryl, long alkyl, silyl, or acetal protective groups (**4a**, **24–26**) also provided the corresponding esters (**27–30**) in high diastereoselectivity.

The obtained phosphonochalcogenoates can be further derivatized (Scheme 2). For example, chiral phosphine **32**, which can be used in nucleophilic asymmetric catalysis,¹⁹ was obtained from dialkylation–deselenation of phosphonoseleenoate **21** with *n*-BuLi. Moreover, the alcoholysis of phosphonate **31** with sodium ethoxide afforded diethyl phosphonate **33**. Under the applied conditions, binaphthol was recovered in both cases.

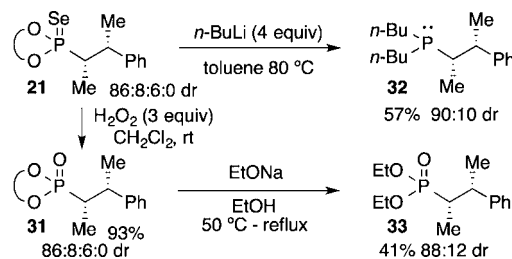
In conclusion, we have demonstrated the diastereoselective formation of chiral carbon centers in binaphthyloxy-substituted phosphonochalcogenoates through a deprotonation–alkylation sequence. The judicious choice of the elements on the phosphorus(V) atom provides control over the reactivity of the phosphorus reactants and thus affords products with tri- and tetrasubstituted chiral carbon centers, which represent key precursors for a range of optically active organophosphorus compounds. Current studies in our laboratories are focused on developing asymmetric reactions with the obtained products as

Table 5. Reaction Scope with Respect to Phosphonates

reactant	EX	product	yield (%) ^a
4a , 24–26		27–30	
4a H Me 7:93 dr	EtI	27 Et Me 5:>95 dr	43
24 H Me 5:>95 dr	EtI	28 Et Me 10:90 dr	60
25 H Me 10:90 dr	EtI	29 Et Me 13:87 dr ^b	59
26 H Me 10:90 dr	PhCH ₂ Br	30 Bn Me 5:>95 dr	85

^aIsolated yield; dr determined by ³¹P NMR analysis of the crude reaction mixtures. ^bDetail of the elucidation of the absolute configuration is shown in the Supporting Information.

Scheme 2. Derivatization of Phosphonoseleenoate **21 and Phosphonate **31****



well as gaining more detailed insights into the stereochemistry and the reaction mechanism.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b02578.

Experimental details and data (PDF)

X-ray data for **6a** (CIF)

X-ray data for **6c** (CIF)

X-ray data for **21** (CIF)

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: mtoshi@gifu-u.ac.jp.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

This work was supported by the JSPS KAKENHI Grant Nos. JP16H01139 (Middle Molecular Strategy) and JP16H04146.

■ REFERENCES

- (1) Demmer, C. S.; Krogsgaard-Larsen, N.; Bunch, L. *Chem. Rev.* **2011**, *111*, 7981.
- (2) (a) Fields, S. C. *Tetrahedron* **1999**, *55*, 12237. (b) Kurz, T.; Schluter, K.; Kaula, U.; Bergmann, B.; Walter, R. D.; Geffken, D. *Bioorg.*

- Med. Chem.* **2006**, *14*, 5121. (c) Andaloussi, M.; Henriksson, L. M.; Wieckowska, A.; Lindh, M.; Bjorkelid, C.; Larsson, A. M.; Suresh, S.; Iyer, H.; Srinivasa, B. R.; Bergfors, T.; Unge, T.; Mowbray, S. L.; Larhed, M.; Jones, T. A.; Karlen, A. *J. Med. Chem.* **2011**, *54*, 4964. (d) Deng, L.; Endo, K.; Kato, M.; Cheng, G.; Yajima, S.; Song, Y. *ACS Med. Chem. Lett.* **2011**, *2*, 165. (f) Ordóñez, M.; Sayago, F. J.; Cativiela, C. *Tetrahedron* **2012**, *68*, 6369. (g) Maji, B.; Yamamoto, H. *Synlett* **2015**, *26*, 1528. (h) Olszewski, T. K.; Majewski, M. *Tetrahedron: Asymmetry* **2015**, *26*, 846. (i) Ordóñez, M.; Viveros-Ceballos, J. L.; Cativiela, C.; Sayago, F. J. *Tetrahedron* **2015**, *71*, 1745.
- (3) (a) Zhao, D.; Mao, L.; Wang, L.; Yang, D.; Wang, R. *Chem. Commun.* **2012**, *48*, 889. (b) Hayashi, M.; Shiomi, N.; Funahashi, Y.; Nakamura, S. *J. Am. Chem. Soc.* **2012**, *134*, 19366. (c) Hatano, M.; Horibe, T.; Ishihara, K. *Angew. Chem., Int. Ed.* **2013**, *52*, 4549. (d) Yin, L.; Bao, Y.; Kumagai, N.; Shibasaki, M. *J. Am. Chem. Soc.* **2013**, *135*, 10338. (e) Li, G.; Wang, L.; Yao, Z.; Xu, F. *Tetrahedron: Asymmetry* **2014**, *25*, 989. (f) Kuchurov, I. V.; Nigmatov, A. G.; Kryuchkova, E. V.; Kostenko, A. A.; Kucherenko, A. S.; Zlotin, S. G. *Green Chem.* **2014**, *16*, 1521. (g) Olszewski, T. K. *Synthesis* **2014**, *46*, 403. (h) Zhang, H.; Sun, Y.-M.; Zhao, Y.; Zhou, Z.-Y.; Wang, J.-P.; Xin, N.; Nie, S.-Z.; Zhao, C.-Q.; Han, L.-B. *Org. Lett.* **2015**, *17*, 142. (i) Roy, S.; Sutivisedsak, N.; Hamper, B. C.; Lyss, A.; Spilling, C. *Synthesis* **2015**, *47*, 3669. (j) Olszewski, T. K. *Tetrahedron: Asymmetry* **2015**, *26*, 393.
- (4) (a) Cheruku, P.; Paptchikhine, A.; Church, T. L.; Andersson, P. G. *J. Am. Chem. Soc.* **2009**, *131*, 8285. (b) Zhang, J.; Dong, K.; Wang, Z.; Ding, K. *Org. Biomol. Chem.* **2012**, *10*, 1598. (c) Dong, K.; Wang, Z.; Ding, K. *J. Am. Chem. Soc.* **2012**, *134*, 12474. (d) Tao, X.; Li, W.; Ma, X.; Li, X.; Fan, W.; Zhu, L.; Xie, X.; Zhang, Z. *J. Org. Chem.* **2012**, *77*, 8401. (5) (a) Nishimura, T.; Hirabayashi, S.; Yasuhara, Y.; Hayashi, T. *J. Am. Chem. Soc.* **2006**, *128*, 2556. (b) Cheng, E. W. C.; Mandalia, R. T.; Motevalli, M.; Mothia, B.; Patanwadia, Y.; Wyatt, P. B. *Tetrahedron* **2006**, *62*, 12398. (c) Guo, Y.-G.; Li, D.-P.; Li, Y.-L.; Wang, H.-M.; Xiao, W.-J. *Chirality* **2009**, *21*, 777. (d) Corbett, M. T.; Johnson, J. S. *J. Am. Chem. Soc.* **2013**, *135*, 594. (e) Duan, S.-W.; Liu, Y.-Y.; Ding, W.; Li, T.-R.; Shi, D.-Q.; Chen, J.-R.; Xiao, W.-J. *Synthesis* **2013**, *45*, 1647. (f) Jang, K. P.; Hutson, G. E.; Johnston, R. C.; McCusker, E. O.; Cheong, P. H.-Y.; Scheidt, K. A. *J. Am. Chem. Soc.* **2014**, *136*, 76. (g) Maji, B.; Yamamoto, H. *Angew. Chem., Int. Ed.* **2014**, *53*, 14472. (h) Chen, D.; Zhu, D.-X.; Xu, M.-H. *J. Am. Chem. Soc.* **2016**, *138*, 1498.
- (6) (a) Ordóñez, M.; Hernández-Fernández, E.; Xahuentitla, J.; Cativiela, C. *Chem. Commun.* **2005**, 1336. (b) Fernández, M. C.; Díaz, A.; Guillín, J. J.; Blanco, O.; Ruiz, M.; Ojea, V. *J. Org. Chem.* **2006**, *71*, 6958. (c) Rai, V.; Mobin, S. M.; Namboothiri, I. N. N. *Tetrahedron: Asymmetry* **2007**, *18*, 2719. (d) Ordóñez, M.; Hernández-Fernández, E.; Rojas-Cabrera, H.; Labastida-Galván, V. *Tetrahedron: Asymmetry* **2008**, *19*, 2767. (e) Mihai, C.; Yue, X.; Zhao, L.; Kravchuk, A.; Tsai, M.-D.; Bruzik, K. S. *New J. Chem.* **2010**, *34*, 925. (f) Reznikov, A. N.; Sibiryakova, A. E.; Rybakov, V. B.; Klimochkin, Y. N. *Tetrahedron: Asymmetry* **2015**, *26*, 1050.
- (7) (a) Denmark, S. E.; Dorow, R. L. *J. Org. Chem.* **1990**, *55*, 5926. (b) Denmark, S. E.; Chen, C.-T. *J. Org. Chem.* **1994**, *59*, 2922. (c) Denmark, S. E.; Chen, C.-T. *J. Am. Chem. Soc.* **1995**, *117*, 11879. (d) Kranz, M.; Denmark, S. E. *J. Org. Chem.* **1995**, *60*, 5867. (e) López, B.; Maestro, A.; Pedrosa, R. *Synthesis* **2006**, *2006*, 817. (f) Denmark, S. E.; Marlin, J. E.; Rajendra, G. *J. Org. Chem.* **2013**, *78*, 66.
- (8) (a) Bennani, Y. L.; Hanessian, S. *Tetrahedron* **1996**, *52*, 13837. (b) Yuan, C.-Y.; Chen, Q.-Y. *Chin. J. Chem.* **2005**, *23*, 1671.
- (9) Balczewski, P.; Szadowiak, A.; Bodzioch, A.; Bialas, T.; Wieczorek, W. M.; Szyrej, M. *J. Organomet. Chem.* **2007**, *692*, 997.
- (10) (a) Sai, M.; Yamamoto, H. *J. Am. Chem. Soc.* **2015**, *137*, 7091. (b) Kim, J. H.; Coric, L.; Palumbo, C.; List, B. *J. Am. Chem. Soc.* **2015**, *137*, 1778. (c) Yan, S.-B.; Zhang, S.; Duan, W.-L. *Org. Lett.* **2015**, *17*, 2458. (d) Yang, B.-M.; Cai, P.-J.; Tu, Y.-Q.; Yu, Z.-X.; Chen, Z.-M.; Wang, S.-H.; Zhang, F.-M. *J. Am. Chem. Soc.* **2015**, *137*, 8344. (e) Senda, Y.; Nakajima, K.; Nishibayashi, Y. *Angew. Chem., Int. Ed.* **2015**, *54*, 4060. (f) Liu, C.-Y.; Han, F.-S. *Chem. Commun.* **2015**, *51*, 11844. (g) Tsui, G. C.; Liu, L.; List, B. *Angew. Chem., Int. Ed.* **2015**, *54*, 7703. (h) Liu, L.; Leutzsch, M.; Zheng, Y.; Alachraf, M. W.; Thiel, W.; List, B. *J. Am. Chem. Soc.* **2015**, *137*, 13268.
- (11) For reviews, see: (a) Akiyama, T. *Chem. Rev.* **2007**, *107*, 5744. (b) Terada, M. *Chem. Commun.* **2008**, 4097. (c) Parmar, D.; Sugiono, E.; Raja, S.; Rueping, M. *Chem. Rev.* **2014**, *114*, 9047.
- (12) (a) Tanaka, K.; Ohta, Y.; Fuji, K. *Tetrahedron Lett.* **1993**, *34*, 4071. (b) Tanaka, K.; Otsubo, K.; Fuji, K. *Tetrahedron Lett.* **1996**, *37*, 3735. (c) Tanaka, K.; Watanabe, T.; Ohta, Y.; Fuji, K. *Tetrahedron Lett.* **1997**, *38*, 8943. (d) Tanaka, K.; Watanabe, T.; Shimamoto, K.; Sahakitpichan, P.; Fuji, K. *Tetrahedron Lett.* **1999**, *40*, 6599. (e) Yamazaki, J.; Watanabe, T.; Tanaka, K. *Tetrahedron: Asymmetry* **2001**, *12*, 669. (f) Bedekar, A. V.; Watanabe, T.; Tanaka, K.; Fuji, K. *Tetrahedron: Asymmetry* **2002**, *13*, 721. (g) Yamazaki, J.; Bedekar, A. V.; Watanabe, T.; Tanaka, K.; Watanabe, J.; Fuji, K. *Tetrahedron: Asymmetry* **2002**, *13*, 729.
- (13) Murai, T.; Maekawa, Y.; Hirai, Y.; Kuwabara, K.; Minoura, M. *RSC Adv.* **2016**, *6*, 15180.
- (14) Murai, T.; Maekawa, Y.; Monzaki, M.; Ando, T.; Maruyama, T. *Chem. Commun.* **2013**, *49*, 9675.
- (15) A difference in reactivity of structurally similar carbanions as a function of the element attached to the phosphorus atom has previously been reported: Mons, S.; Sabourault, N.; Klein, E.; Mioskowski, C.; Lebeau, L. *Tetrahedron Lett.* **2001**, *42*, 7547.
- (16) The stereochemistry of some of the major products depicted with tapered hashed lines was assigned by analogy of their NMR spectra with those of **6a**, **6c**, and **21**. The origin of high diastereoselectivity is speculated with the stereochemical model. For details, see the [Supporting Information](#).
- (17) For aminophosphonates having tetrastituted carbon centers, see: (a) Bera, K.; Namboothiri, I. N. N. *Asian J. Org. Chem.* **2014**, *3*, 1234. (b) Boduszek, B.; Olszewski, T. *Arkivoc* **2017**, 173 and refs cited therein.
- (18) (a) Delarue-Cochin, S.; Pan, J.-J.; Dauteloup, A.; Hendra, F.; Angoh, R. G.; Joseph, D.; Stephens, P. J.; Cave, C. *Tetrahedron: Asymmetry* **2007**, *18*, 685. (b) Shibata, M.; Ikeda, M.; Motoyama, K.; Miyake, Y.; Nishibayashi, Y. *Chem. Commun.* **2012**, *48*, 9528. (c) Bau, S., Jr.; Chen, I.-C.; Yanga, Z.-X.; Zhu, J.-L. *J. Chin. Chem. Soc.* **2013**, *60*, 531.
- (19) For reviews, see: (a) Fan, Y. C.; Kwon, O. *Chem. Commun.* **2013**, *49*, 11588. (b) Xiao, Y.; Sun, Z.; Guo, H.; Kwon, O. *Beilstein J. Org. Chem.* **2014**, *10*, 2089.